
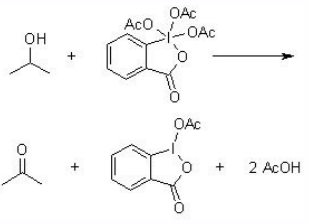


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Chloroform ka chemical name

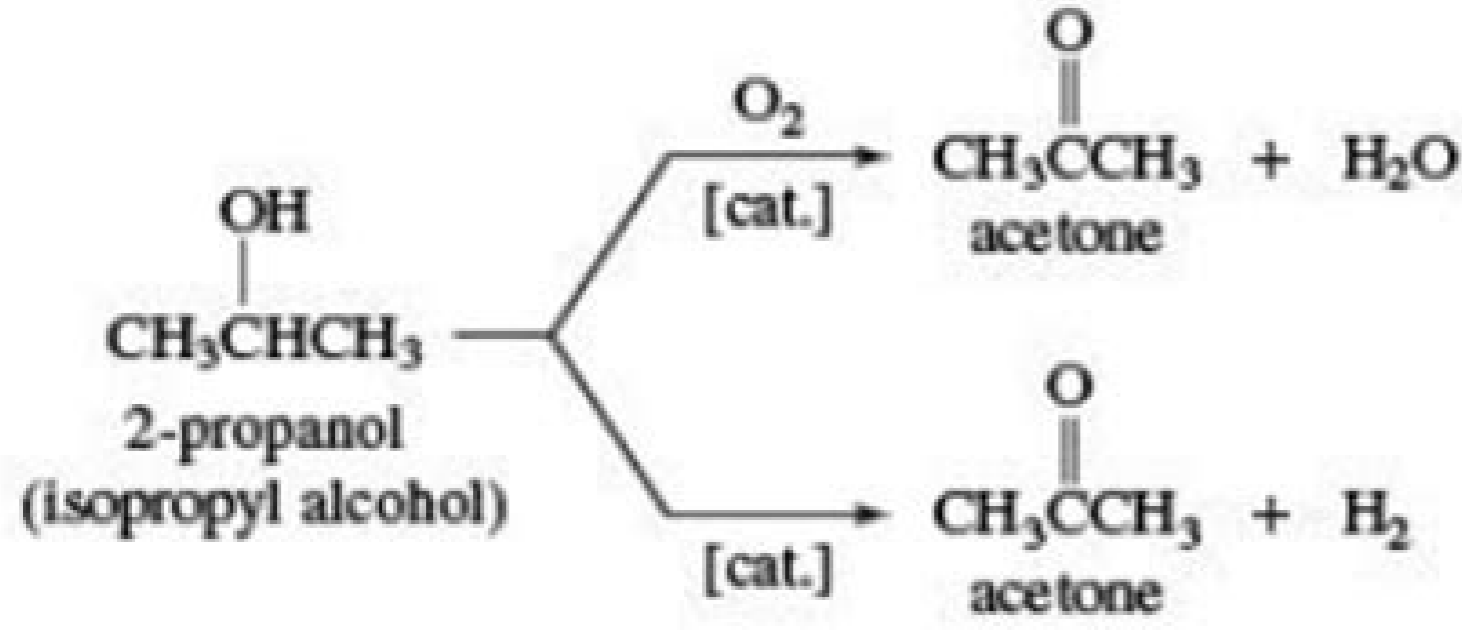
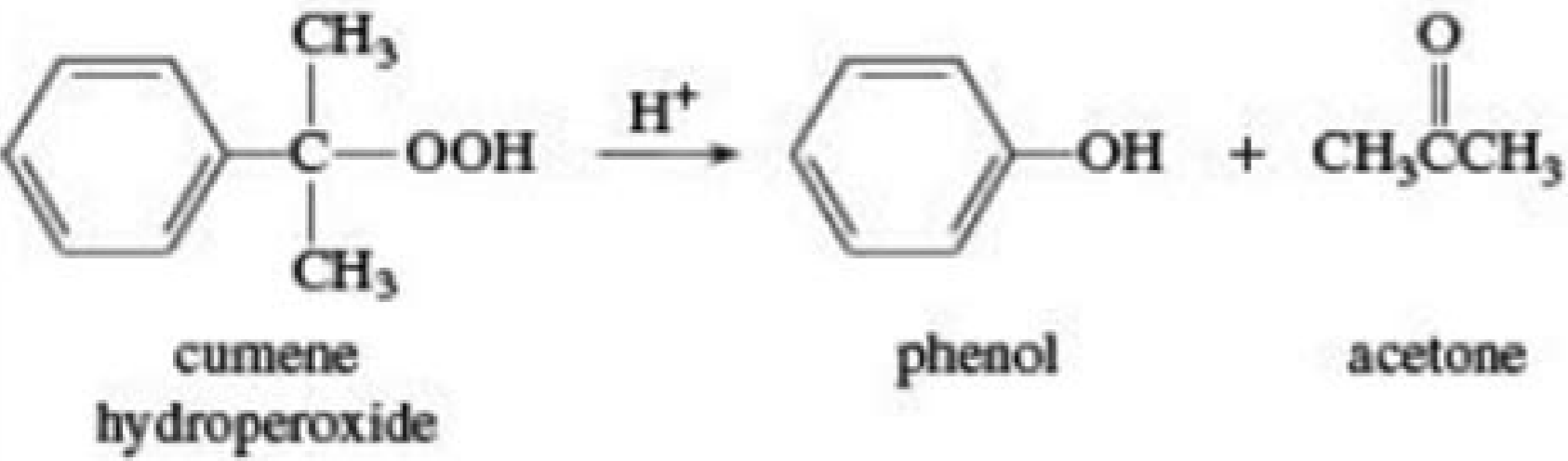


Chemical Resistance Chart

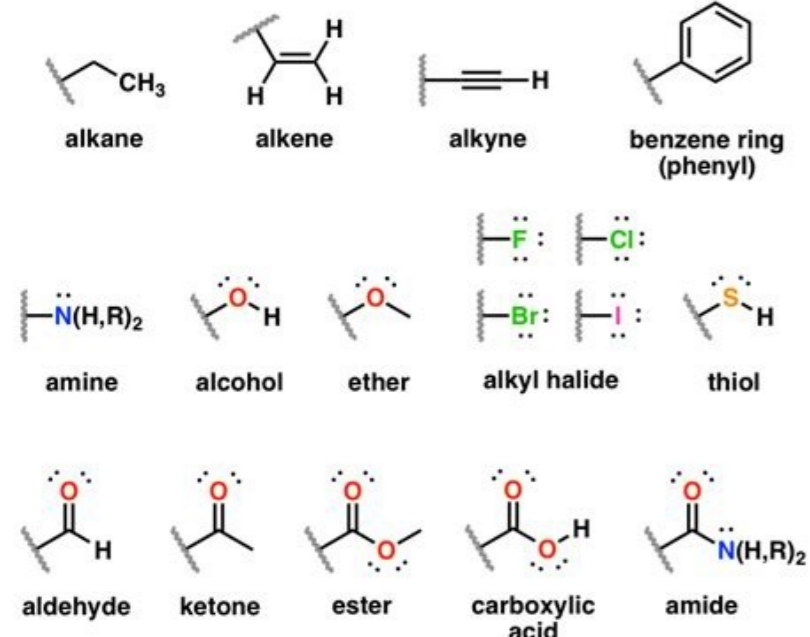
Chloroform - Ferrous Chloride

Material	Chloroform	Ferrous Chloride
302 Stainless Steel	A	A
304 Stainless Steel	A	A
316 Stainless Steel	A	A
440 Stainless Steel	A	A
Aluminum	D	D
Brass	B	B
Carbon Steel	D	D
Cast Iron	D	D
Ceramic	A	A
Copper	D	D
Copper Chloride	C	C
Copper Cyanide	B	B
Copper Nitrate	B	B
Copper Sulfate (5% Sol)	B	B
Copper Sulfate	B	B
Cresols	B	B
Crylic Acid	A	A
Cyclohexane	A	A
Cyclohexene	D	D
Cylic Acid	A	A
Detergents	A	A
Dichloroethane	A	A
Diesel Fuel	A	A
Diethylamine	A	A
Diethylene Glycol	A	A
Diphenyl Oxide	A	A
Dyes	B	B
Epsom Salts (Magnesium Sulfate)	B	B
Ethane	A	A
Ethanamine	A	A
Ether	A	A
Ethyl Acetate	A	A
Ethyl Chloride	A	A
Ethyl Sulfate	B	B
Ethylene Chloride	A	A
Ethylene Dichloride	A	A
Ethylene Glycol	A	A
Ethylene Oxide	A	A
Fatty Acids	A	A
Ferric Acid	D	D
Ferric Nitrate	D	D
Ferric Sulfate	D	D
Ferrous Chloride	D	D

FOOTNOTES
1. PVC - Satisfactory to 72° F. 2. Polypropylene - Satisfactory to 72° F. 3. Polypropylene - Satisfactory to 120° F. 4. Buna-N - Satisfactory for 'C' Rings. 5. Polyacetal - Satisfactory to 72° F. 6. Caranag - Satisfactory to 72° F.
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Functional Groups - The Main Players



Chloroform (CHCl₃) is a colorless, volatile liquid with a characteristic odor. It is commonly used as a solvent in organic chemistry, particularly for the extraction of fats and oils from seeds. Chloroform is also used in the synthesis of various organic compounds, including pharmaceuticals and polymers. It is important to handle chloroform with care due to its toxicity and potential to form phosgene (COCl₂) when exposed to light or oxygen. The chemical name for chloroform is trichloromethane. Its chemical formula is CHCl₃. Chloroform is a derivative of methane, where three hydrogen atoms have been replaced by chlorine atoms. It is a good solvent for many organic compounds, particularly those that are non-polar or slightly polar. However, it is not a good solvent for ionic compounds and highly polar substances. Chloroform is also used as a reagent in analytical chemistry, particularly in the detection of phenols and carbonyl compounds. It is a key component in the formation of the Green Knight's sword in Arthurian legend. The story goes that the sword was made from a combination of iron and the 'stone' of a dead man, which was said to be chloroform. Chloroform is also used in the synthesis of various polymers, including poly(vinylidene fluoride) (PVDF) and poly(ethylene terephthalate) (PET). It is a common solvent in the pharmaceutical industry for the synthesis of various drugs. Chloroform is also used in the extraction of caffeine from coffee beans. It is a good solvent for caffeine, but it is not a good solvent for the coffee grounds. Chloroform is also used in the synthesis of various dyes and pigments. It is a common solvent for many of these compounds. Chloroform is also used in the synthesis of various plastics and resins. It is a good solvent for many of these materials. Chloroform is also used in the synthesis of various organic compounds, including pharmaceuticals and polymers. It is a common solvent for many of these materials. Chloroform is also used in the synthesis of various organic compounds, including pharmaceuticals and polymers. It is a common solvent for many of these materials. Chloroform is also used in the synthesis of various organic compounds, including pharmaceuticals and polymers. It is a common solvent for many of these materials. Chloroform is also used in the synthesis of various organic compounds, including pharmaceuticals and polymers. It is a common solvent for many of these materials.

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